Synthesis of 4-phenylazo-1-naphthol and its antifungal activity against *Fusarium oxysporum* f. sp. *lycopersici*

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Abstract

This study is based on the synthesis of 4-phenylazo-1-naphthol ($C_{16}H_{12}N_2O$) as an azo coupled dye through a coupling reaction of phenyl diazonium salt and α - naphthol in ice-cold chilled water. Azo coupling involved an electrophilic substitution reaction of phenyl diazonium cation with α - naphtholate ion, the coupling partner. The 4-phenylazo-1-naphthol was characterized through fourier transform infrared spectroscopy (FTIR). Antifungal activity of 4-phenyl-azo-1-naphthol was checked against *Fusarium oxysporum* f. sp. *lycopersici* (FOL), a soil-borne fungal pathogen causing wilt disease in tomato. Eight concentrations ranging from 0.78 to 100 mg mL⁻¹ were tested against the fungus. None of the concentration suppressed the fungal growth. Instead, all the concentrations variably enhanced the fungal biomass over control by 8–28%. This study concludes that the synthesized compound did not possess antifungal potential against FOL.

Keywords: Antifungal agent, Azo dye, Fusarium wilt, Synthesis

Introduction

Fusarium species are ubiquitous in nature and cause soil-borne diseases of great economic importance in horticultural and food crops worldwide (Bodah, 2017; Javaid et al., 2018; Akhtar et al., 2020). In addition to the losses caused by Fusarium species, some of them are capable of producing mycotoxins in food and agricultural commodities (Gautier et al., 2020). F. oxysporum f. sp. lycopersici is one of the most prevalent fungal pathogens that induce vascular wilt of tomatoes (Khurshid et al., 2014; Devi et al., 2022). It is a major soil-borne systemic disease occurring throughout tomato-growing areas and severely affects crops (Kamali et al., 2022). In the absence of a host, FOL can survive indefinitely in infected soils as dormant propagules. The presence of host roots triggers the chlamydospore's germination and infection hyphae adhere to the root surface for penetration. Later on, it infects vascular tissues and inhibits the plant xylem vessels, resulting in vessel clogging, and the appearance of wilt-like symptoms (Srinivas et al., 2019). It displays a unique pathway of infection where fungal mycelium invades intercellularly root cortical cells and enters inside xylem vessels. The pathogen produces microconidia within these vessels and is transported to upper vessels eventually causing the host plant's death

(Maurya et al., 2019).

Several disease management strategies such as crop rotation, resistant cultivars, cultural techniques, and biological and chemical control are available against FOL (Javaid and Bashir, 2015; Khurshid et al., 2017; Maurya et al., 2019). Among them, chemical control is an effective measure of controlling FOL (Akram et al., 2018). The application of fungicides namely benzimidazoles, fuberidazole, thiophante, benomyl, carbendazim, thiabendazole, propiconazole and prochloraz have provided substantial control of Fusarium wilt, root rot and crown rot of tomatoes (Nel et al., 2007). Mohamed et al. (1980) reported that the application of Vitavax-captan or thiuram as fungicidal treatment is effective in controlling the growth of the pathogen. Copper oxychloride and metamidoxime have also been found effective against FOL as new products (Nedelcu and Alexandri, 1995). In addition, azoxystrobin, iprobenfos, tebuconazole, streptomycin, fentin hydroxide, carbendazim, strobilurin, epoxiconazole, myclobutanil, prochloraz, fosetyl-al, flumorph, bromuconazole, benlate, fludioxonil and pyrimethanil have been reported very effective against FOL (Akram et al., 2018). However, there is need to explore more and more compounds against Fusarium species.

Azo dyes are organic compounds containing R-N=N-R' group where R and R' are generally aryl

groups. Prior to the synthesis of Mauvaine in 1856, the first synthetic organic dye (Filarowski, 2010), roots and leaves of some plant species were used to dye the textile (Tsemeugne et al., 2017). Nowadays, almost all the molecules used in dyeing industry are synthetic, mostly azo dyes (Bae and Freeman, 2007; Rathod and Thakr, 2013). In addition, azo dyes are their pharmaceutical and also known for antimicrobial properties (Sopbué et al., 2013; Gaffer et al., 2016). Therefore, the present investigation was carried out to synthesize 4-phenylazo-1-naphthol and evaluation of its antifungal potential against F. oxvsporum f. sp. lvcopersici.

Materials and Methods

Analytical grade aniline (93.13 g mol⁻¹, pure yellowish liquid, b.p. 184 °C, flash point 158 °F), α -naphthol (m.p. 95 to 96 °C, \geq 99.99% pure), sodium nitrite (\geq 99.99% pure), hydrochloric acid and sodium hydroxide were procured from central chemicals and used without further treatment.

Synthesis of 4-phenylazo-1-naphthol

Five grams of aniline were dissolved in 16 mL of conc. HCl and the same amount of water in a small beaker. The diazotization step was carried out by adding a solution of 4 g of sodium nitrite in 20 mL of water. Thereafter, a solution of 7.8 g of 1naphthol in 45 mL of 10% sodium hydroxide was prepared in a beaker. The solution was cooled to 5 °C by immersing in an ice bath ice. The 1-naphthol solution was vigorously stirred followed by slow addition of cold diazonium salt solution developing a dark brown color which later on converted to dark brown crystals of 2-phenylzo 1-naphthol. The mixture was allowed to stand in an ice bath for 30 min with occasional stirring. Filtered the mixture through a funnel, spread it on filter paper with a glass rod, let it dry in indirect sunlight, then put the powdery granules in a vial.

Antifungal activity

To prepare a stock solution of 100 mg mL⁻¹ of the synthesized compound, 0.6 g were dissolved in 0.5 mL dimethyl sulfoxide (DMSO) and raised the volume to 6 mL by adding autoclaved malt extract broth (MEB). It was serially double diluted to prepare lower concentrations of 50, 25, 12.5, 6.25, 3.12, 1.56 and 0.78 mg mL⁻¹. A similar series of concentrations of control treatments were prepared with the same amount of DMSO but without the compound. To each test tube (5-mL), 1.0 mL of the growth media of different concentrations was poured. Culture of F. oxysporum f. sp. lycopersici was obtained from Biofertilizers and Biopesticides Lab., Punjab University Lahore (Fig. 1A). and inoculated with 20 µL of S. rolfsii was added. Each treatment was replicated three times. Tubes were incubated at 28 °C for 7 days. Thereafter, fungal biomass was collected, dried and weighed (Javaid

and Samad, 2012).

Statistical analysis

All the data were analyzed by ANOVA followed by application of LSD test at 5% level of significance using Statistix 8.1.

Results and Discussion

Physical characteristics of the synthesized compound

Fig. 1 B shows the physical appearance of synthesized azo dye 4-phenyl-azo-1-naphthol. The product appeared as crystalline solid with sharp melting point around $165 \, {}^{\circ}\text{C}$.

FTIR spectra of the synthesized compound

The azo coupling preferably occurs at the para position of the same ring since the charge density gets reinforced at the para position. It involves an electrophilic substitution reaction of phenyl diazonium cation with the coupling partner α naphtholate ion. Fig. 2 describes the proposed route for the synthesis of 4-phenylazo-1-naphthol while Fig. 3 displays the FTIR spectrum of the prepared 4phenylazo-1-naphthol. The broad peak at 3550-3215 cm⁻¹ shows the presence of H-bonded hydroxyl (-OH) groups, peaks at 2900–2950 cm⁻¹ correspond to sp2 hybridized methine (-CH) groups of Ph-CH. A sharp peak at the 1510-1500 cm⁻¹ shows the azo group (-N=N-) in 4-phenylazo-1-naphthol. Benzene ring was confirmed by the presence of different peaks around 1620-1680 cm⁻¹. The (C-O-C) stretching vibrations were around 1016 cm⁻¹. The characteristic peak 1450–1400 cm⁻¹ shows the bending frequency of -NH- group under the influence of azo coupling groups. The main characteristic peaks and their assignments are given in Table 1.

Antifungal activity of the synthesized compound

The synthesized azo dye compound, 4phenyl-azo-1-naphthol did not show antifungal activity against S. rolfsii. Instead, all the concentrations of this compound variably increased fungal biomass by 8-28% over control (Fig. 4). Kovac et al. (2014) reported that azo compounds are useful and perform different mechanisms of action. These compounds are importance due to their versatile biological activities and functions. In contrast to the findings of the present study, many azo dye compounds have been reported as antifungal in nature (Raghavendra and Kumar, 2015). By contrast, some studies also show that noneffectiveness of azo compounds against fungal growth. Prashantha et al. (2021) worked on novel derivatives of azo dyes (6–9) that were prepared by amino-methylbenzoic acid diazotization and tested them against Fusarium oxysporum. The findings were in accordance with present studies where dyes

6 and 8 exhibited moderate antifungal activities, whereas dyes 7 and 9 were not effective. Similarly, some new azo dyes were synthesized by trichlorotriazine moieties and pyrazole and their antimicrobial potential was evaluated against Candida albicans (Rizk et al., 2015). One series of dyes significantly inhibited the mycelial growth of C. albicans while the rest of tested compounds were not effective against the fungal pathogen. Likewise, Singh et al. (2014) prepared a series of novel azo evaluated them dves and against С. albicans, Aspergillus flavus and A. niger. Among the tested compounds only two were found effective against the tested fungal species. The same findings

were given by Kumar *et al.* (2013) where only one compound showed antifungal potential while rest of the dyes were not antifungal in nature.

Conclusion

This study concludes that the synthesized compound 4-phenylazo-1-naphthol did not show antifungal activity against *S. rolfsii.* Further studies are suggested to check antifungal activity of this compound against other phytopathogens. In addition, derivatives of this compound can be prepared to enhance its antifungal properties.

Table 1: The main characteristic IR absorption peaks in 4-phenylazo-1-naphthol and their assignments.

4-phenylazo-1-naphthol	Assignments
b 3600-3250 cm ⁻¹	vO-H stretching
m 3050-2850 cm ⁻¹	vC-H stretching (sp ² and sp ³)
m 1614 cm ⁻¹	vC-C stretching
vs 1549 cm ⁻¹	vCC + vCH naphthol
m 1478-1531 cm ⁻¹	vCC + vCH + vCN
b 1406 cm ⁻¹	vCO + v CH + v CN + v CC
teath like 1478-1531 cm ⁻¹	aromatic $vC=C$
s 1215 cm ⁻¹	bending C-N
s 1504-1530 cm ⁻¹	vN=N stretching
s 1000-1015 cm ⁻¹	vC-O stretching
s 757 cm ⁻¹	wCH naphthol

Abbreviations: vs: very strong, w: weak and m: medium.



Fig. 1: Pure culture of *Fusarium oxysporum* f. sp. *lycopersici* (A) and synthesized azo dye 4-phenyl-azo-1-naphthol (B).



4-[(E)-phenyldiazenyl]naphthalen-1-ol

Fig. 2: Proposed route for the synthesis of 4-phenylazo-1-naphthol.



Fig. 3: FTIR spectrum of 4-phenylazo-1-naphthol.



Fig. 4: Effect of different concentrations of azo dye on the growth of Fusarium oxysporum. Vertical bars show standard errors of means of three replicates. Values with different letters at their top show significant difference ($P \le 0.05$) as determined by LSD Test.

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